

GPS Safety Summary

Substance Name:

Di-2-ethylhexyl peroxydicarbonate

1. General Statement

The substance is primarily used as a radical initiator. The substance is used at industrial stage and is not sold to consumers.

2. Chemical Identity

Name: Brand names:	Di-2-ethylhexyl peroxydicarbonate Luperox® 223S, Luperox® 223MO75, Luperox® 223M75, Luperox® 223V75, Luperox® 223EN grades		
Chemical name (IUPAC):	3,3'-[dioxybis(carbonyloxymethylene)]diheptane		
CAS number(s):	16111-62-9		
EC number:	240-282-4		
Molecular formula:	$C_{18}H_{34}O_6$		
Structure:			
Bu			

3. Use and applications

The substance is primarily used as a radical initiator to induce polymerisation (free radical polymerisation process) for the manufacture of plastic (PVC).

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4. Physical / Chemical properties

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Organic peroxides are thermally unstable substances or mixtures, which can undergo exothermic self-accelerating decomposition.

Di-2-ethylhexyl peroxydicarbonate is classified as Organic Peroxide, Type C, H242 (see section 8 and 9 below).

Property	Value
Physical state	Liquid at 20°C and 1013 hPa
Colour	Colourless
Odour	Unpleasant
Molecular weight	346.46 g/mol
Density	0.9 g/cm ³ at 0°C
Vapour pressure	<0.001 Pa at 25°C
Freezing point	<-70°C
Boiling point	Not relevant, the substance decomposes before boiling
Self-Accelerating Decomposition Temperature (SADT)	SADT varies with concentration and type of diluent. The lowest SADT for the most concentrated product (75%) is 5°C
Explosive properties	Risk of explosion if heated under confinement
Water solubility	Not relevant, instable in water (rapid hydrolysis in especially 2-ethylhexanol)
Octanol-water partition coefficient (LogKow)	6.97 at 25°C

5. Health Effects

The substance has a low toxicity although there is some concern regarding the irritation potential.

Effect Assessment	Result
Acute Toxicity Oral / inhalation / dermal	No acute toxic effects following oral / dermal exposures. No data available following acute inhalation exposure.
Irritation / corrosion Skin / eye	Causes skin irritation. May cause serious eye damage.
Sensitisation	May cause skin allergic reaction.
Toxicity after repeated exposure Oral / inhalation / dermal	Does not cause significant target toxicity following oral exposure.
Genotoxicity / Mutagenicity	The substance has no <i>in vitro</i> genotoxic potential.
Carcinogenicity	No data available.
Reproductive / Developmental Toxicity	No data available.

6. Environmental Effects

Bis(2-ethylhexyl) peroxydicarbonate undergoes immediate hydrolytical degradation in contact with water. Any possibility for exposure of aquatic organisms to bis(2-ethylhexyl) peroxydicarbonate can therefore be excluded, only degradation products are considered. Lack of persistence and bioaccumulation potential indicate that the substance is neither PBT nor vPvB.

Effect Assessment	Result
Aquatic Toxicity	Toxic to daphnias

Fate and behaviour	Result
Biodegradation	Readily biodegradable, but failing the 10-d windows
Bioaccumulation potential	No potential to bioaccumulate
PBT / vPvB conclusion	Not considered as PBT or vPvB

7. Exposure

7.1 Human health

The manufacture of organic peroxide is a closed process that occurs behind anti-deflagration walls, which minimizes worker exposure during the production process.

However, workers can be exposed during loading/unloading operations, mixing, sampling or maintenance operations.

The primary routes of industrial/professional exposure of the substance are skin contact and inhalation.

In addition, the general population is not expected to be exposed to this organic peroxide by inhalation, dermal or oral exposure, as the product does not remain in the plastic products.

Based on the risk assessment, risk is controlled when activities are carried out under conditions recommended in the extended safety data sheet (chapter 8 and exposure scenarios)

7.2 Environment

Releases of organic peroxides into the environment are to be expected during production, processing (formulation) and industrial uses mainly via wastewater and lesser amounts via emissions of vapour (due to its physical state and its vapour pressure).

Potential release during production is treated by on-site and off-site risk management measures.

The substance is used for production of polymers, which can either be dry or wet processes. In case of wet processes, releases to water have to be directed to wastewater treatment plant.

The substance is used in low quantity in polymer production, and is almost totally consumed during the process. Therefore the release to environment is very low. Furthermore, the substance is unstable in water and is rapidly decomposed in product of low concern for environment. The use is considered as safe for the environment (this was confirmed by a quantitative risk assessment performed in the framework of REACH regulation)

Human health measures		
Eye/Face protection	Safety glasses/goggles Half-mask during the discharge	
Skin protection	Protective suit	
Hand protection	Gloves: PVC, neoprene (suitable gloves tested to EN374)	

8. Risk Management recommendations

Respiratory protection	Suitable respiratory equipment in case of insufficient ventilation			
Organizational measures	Ensure workers are duly trained to minimize exposure			
Engineering control	Provide sufficient air exchange and/or exhaust ventilation in work rooms			

Environmental measures

Can be discharged in waste water, when in compliance with local regulations. Do not spread sludge on natural soils.

Eliminate the product by incineration after dilution in a suitable flammable solvent (in accordance with local and national regulations) – amount of active oxygen must be below 1%.

Storage and handling

Strictly limit the quantities of product in the work area to those which are absolutely necessary for the work in hand. Great cleanliness in work areas is a necessary and important factor for safety. Never weigh out in the storage room. Handle and open container with care (risk of over pressurization in containers). Eliminate all sources of ignition, and do not generate flames or sparks. Take precautionary measures against static discharges. Apply earthing when transferring from one container to another. Confinement must be avoided. Use explosion protected equipment.

Use non-sparking tools in areas where explosive vapor/air mixtures may occur. Keep product and emptied container away from heat and sources of ignition. Do not cut or weld on or near this container even when empty. Keep away from incompatible materials such as: strong oxidizing agents, powerful reducers, acids, bases, amines, transition metal salts, sulphur compounds, rust, ash, dusts (risk of self-accelerating exothermic decomposition)

Never return any product to the container from which it was originally removed (risk of decomposition).

For the most concentrated products (75% and pure substance): avoid temperatures above -15 °C (prevention of fire and explosion). Storage buildings must be built and equipped so as not to exceed the maximum proscribed temperature limit.

9. Regulatory Information / Classification and Labelling

9.1 Regulatory Information

This substance has been registered under:

- EU Regulation EC 1907/2006 (REACH)

As organic peroxides are sensitive substances (as they are liable to exothermic decomposition), the carriage of tert-amyl hydroperoxide is strongly regulated, under the rules and conditions of class 5.2 of UN Recommendations on the Transport of Dangerous Goods regulation.

%	Form	UN Number	Classification	OP Category
>77- 100%	liquid	UN 3113	OP Type C, Liquid, temperature controlled	Type C: the mixture possesses explosive properties but cannot detonate or deflagrate rapidly or undergo a thermal explosion when packaged

≤77%	liquid	UN 3115	OP Type D, Liquid, temperature controlled	Type D: (i) detonates partially, does not deflagrate rapidly, no violent effects when heated under confinement; or (ii) does not detonate at all, deflagrates slowly, no violent effects when heated under confinement; or (iii) does not detonate or deflagrate, medium effect when heated under confinement
≤62%	stable dispersi on in water	UN 3119	OP Type F, Liquid, temperature controlled	Type F: neither detonates in the cavitated state nor deflagrates at all and shows only a low or no effect when heated under confinement as well as low or no explosive power

9.2 Classification and labelling

Under GHS substances are classified according to their physical, health, and environmental hazards. The hazards are communicated via specific labels and the e-SDS. GHS attempts to standardize hazard communication so that the intended audience (workers, consumers, transport workers, and emergency responders) can better understand the hazards of the chemicals in use. Substances registered for REACH are classified according CLP (EC) 1272/2008, implementation of the GHS in the European Union.

Classification			
 Organic peroxide; type C; Heating may cause a fire. 			
 Skin irritation; Category 2; Ca 	uses skin irritation.		
 Eye Damage; Category 1; Ca 	uses serious eye damage.		
 Skin sensitisation; Category 1 	; May cause an allergic skin reaction		
	Pictogram		
— GHS02: Flame			
 GHS07: Exclamation mark 			
Hazard statement			
 H242: Heating may cause a f 	 H242: Heating may cause a fire. 		
 H315: Causes skin irritation. 			
 H317: May cause an allergic skin reaction. 			
 H318: Causes serious eye damage. 			
Alternative classification according to Globally Harmonized System (GHS)			
 Acute aquatic toxicity; Category 2; Toxic to aquatic life. 			

10. Contact Information within Company

For further information on this substance or product safety summary in general, please contact:

- arkema.peroxides-reach-uses@arkema.com
- ICCA portal where the GPS Safety Summary is posted: <u>http://www.icca-chem.org/en/Home/ICCA-initiatives/global-product-strategy/</u>

11. Date of Issues / Revision

- Date of issue: 2013/11/30
- Date of revision:

12. Disclaimer

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